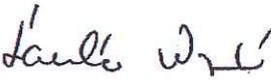
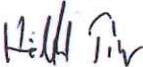


## CELSA - Collaborative research project - Application form - COVER PAGE

1. Identification of the principal investigator of the CELSA application – co-ordinator of the CELSA research project (from partner university OR KU Leuven)
Full name: Ewald Janssens
Faculty/Department: Faculty of Science, Department of Physics and Astronomy
Research unit within Faculty/Department: Laboratory of Solid State Physics and Magnetism
Address: Celestijnenlaan 200 D, B-3001 Leuven, Belgium
University: KU Leuven
Tel: + 32-16-327207
Fax : +32-16-327983
email : <a href="mailto:ewald.janssens@kuleuven.be">ewald.janssens@kuleuven.be</a>
Signature: 

2. Identification of the second investigator
Full name: László Nyulászi
Faculty/Department: Department of Inorganic and Analytical Chemistry
Research unit within Faculty/Department:
Address: Szt Gellert ter 4, H-1521 Budapest, Hungary
University: Budapest University of Technology and Economics
Tel: +36-1-463-3281
email : <a href="mailto:nyulaszi@mail.bme.hu">nyulaszi@mail.bme.hu</a>
Signature: 

3. Identification of the third investigator
Full name: Tibor Höltzl
Faculty/Department: Department of Inorganic and Analytical Chemistry
Research unit within Faculty/Department:
Address: Szt Gellert ter 4, H-1521 Budapest, Hungary
University: Budapest University of Technology and Economics
Tel: +32-20-955-4884
email : <a href="mailto:tibor.holtzl@mail.bme.hu">tibor.holtzl@mail.bme.hu</a>
Signature: 

#### 4. Non confidential and public friendly summary (max. 2000 characters)

**Project title:** Copper clusters as model catalysts for carbon dioxide hydrogenation to methanol

**Summary:**

Carbon capture and utilization is a promising strategy to avoid CO<sub>2</sub> release in the atmosphere. Industrially, CO<sub>2</sub> is converted to the alternative fuel methanol using as catalyst a mixture of Cu, ZnO and Al<sub>2</sub>O<sub>3</sub> nanoparticles. Despite the industrial importance, the nature of the active centers and the working principle of the catalyst are not fully understood on an atomistic level. This lack of knowledge hampers the development of a more effective catalyst that allows hydrogenating CO<sub>2</sub> to methanol at lower temperatures, thereby reducing the energy input and the cost of the process.

In this collaborative research project, we model the active sites of the catalyst by pure and zinc doped copper clusters, either isolated in the gas phase or deposited on an alumina support. The objectives of this novel approach are to measure, with atomic precision, the cluster size dependence of elementary reaction steps and to investigate the role of zinc as promotor element. In a first step, the active sites and the binding configuration of CO<sub>2</sub> and H<sub>2</sub> on isolated pure and Zn-doped copper clusters will be identified. Next, the reaction rates for the adsorption of either CO<sub>2</sub> or H<sub>2</sub> and for the co-adsorption of both molecules on the isolated clusters will be measured. Last, the role of the alumina support on the hydrogenation of CO<sub>2</sub> will be studied for soft-landed clusters. The experimental characterization (Leuven) is complemented with first principles modelling (Budapest) to obtain insight into the structural and electronic effects of the reaction steps.

The project will yield a thorough understanding of the CO<sub>2</sub> hydrogenation, which eventually will allow for the design of an improved catalyst. The project also will strengthen the collaboration between the KU Leuven and the Budapest University of Technology and Economics and will lay the foundations for joint European project applications that make use of clusters as model systems for heterogeneous catalysis.

#### 5. List 5 key words

carbon fixation, gas phase reactivity, clusters of atoms, heterogeneous catalysis, density functional theory calculations